

Notes 10/24

Wednesday, October 24, 2007
9:59 AM



Notes 1024

Audio recording started: 10:00 AM Wednesday, October 24, 2007

Enzymes: Introduction, Rates of Reactions

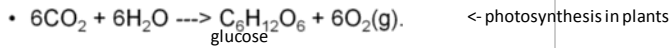
Oct. 24, 2007

Intro to Enzymes

- What is an enzyme?
- Properties
- Classes (Nomenclature)
- Coenzymes

What are Enzymes?

- Proteins (sometimes RNA)
- Catalyze metabolic rxns
- For example: Oxidized form of carbon to more reduced form



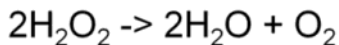
- Living systems use enzymes to accelerate and control the rates of vitally important rxns

- Organisms must be able to catalyze reactions efficiently and selectively

- Catalytic activity depends on native conformation of protein.
- Catalytic activity is lost if protein is denatured or degraded.
- Primary, secondary, tertiary, and quaternary are important for activity.

More examples

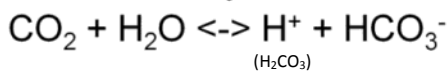
Catalase



Catalase

- Peroxisomes of aerobic cells
- Protects cells from H_2O_2
- Catalase is a tetramer (4 polypeptides)
- 1 catalase can break 40million times in 1 second

Carbonic anhydrase



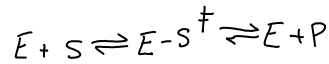
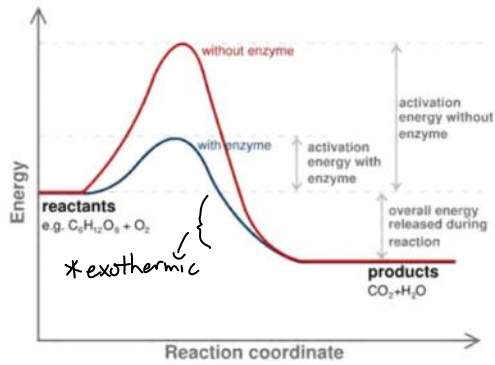
Carbonic anhydrase

- Monomer; metalloenzyme (Zinc 2^+)
- Important in stomach, pancreas, kidney, red blood cells
- Converts CO_2 gas to carbonic acid which can travel through blood
- 1 carbonic anhydrase can process 1million reactions per second

Properties of Enzymes

1. Higher rxn rates
2. Milder rxn conditions
 - physiological pH and temperature
3. Greater rxn specificity
 - Recognize specific reactants and make specific products
 - High degree of stereospecificity
4. Capacity for regulation
 - Regulate enzyme amount and reactivity
 - Regulate by phosphorylation, glycosylation

Free energy diagram



E → enzyme
S → substrate
P → product

Thermodynamics of the system remains the same with or without catalyst. Same energy starting point and same energy needed/released.

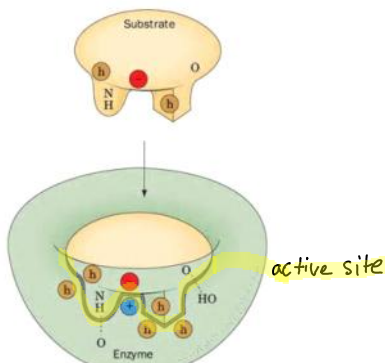


Figure 13-1 An enzyme-substrate complex

- Active site:
 - Pocket
 - Surface is lined with amino acids that react with the substrate
 - Pocket provides what we call a microenvironment that favors a reaction energetically

Need to know these:

Classes of Enzymes

- Enzymes are classified based on the reaction they catalyze.
- Enzymes catalyze reactions in either direction depending on which is thermodynamically favored

1. Oxidoreductases--- oxidation-reduction reactions

ex: Catalase

2. Transferases-- transfers functional groups.

3. Hydrolases--hydrolysis reactions

ex: protease

4. Lyases-- group elimination to change a single bond to a double bond

5. Isomerases-- isomerization

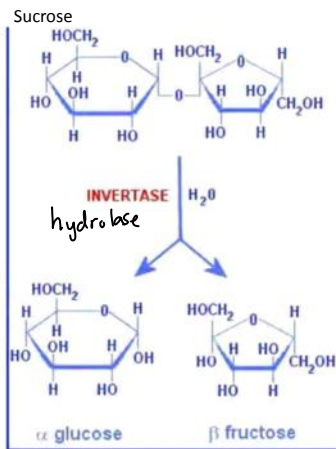
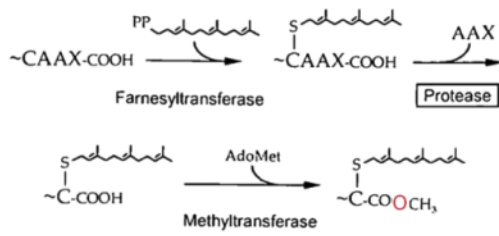
6. Ligases-- bond formation coupled to ATP hydrolysis

4 levels of classification (by Enzyme Commission E.C.)

Catalase E.C.1.11.1.6

*just need to know that this naming exist.

Correction from 10/23 Notes:

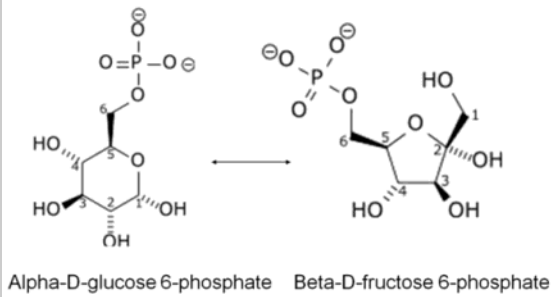


- Invertase is a hydrolase.
 - More technical name is beta-fructosidase
- "inverted" sugar refers to broken down sucrose of glucose and fructose because when broken down it becomes a syrupy mixture (bees do this naturally to make honey)
- Glycosidase
- Used in industry
- Yeast
- Humans have sucrase in the small intestine
- To detect sucrose, use invertase to break down then Tollen's Reagent or Benedicts solution to detect glucose.



Use invertase to make gooey centers in candy, chocolate, etc.

Glucose isomerase



- Enzymes sometimes work by themselves, other times need "sidekick" called a cofactor
- Cofactor can be
 - Inorganic ion (Fe^{2+} , Mg^{2+} , Mn^{2+} , Zn^{2+})
 - Coenzyme - organic or organometallic molecule
 - When cofactor is tightly or covalently bonded to protein it is considered a prosthetic group
 - Carbonic anhydrase uses Zn^{2+}
 - Holoenzyme
 - Protein part is apoenzyme or apoprotein
- Lipid Bilayers
 - Fluid Mosaic Model... plasma membrane is fluid, not static
 - Lateral diffusion
 - Transverse diffusion (flip-flop)
 - Rare
 - Dive into hydrophobic region to go to other side